Received 24 March 2005

Accepted 14 April 2005

Online 23 April 2005

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 293 KMean σ (C–C) = 0.004 Å R factor = 0.061 wR factor = 0.182 Data-to-parameter ratio = 11.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

3,5-Dihydroxy-2-hydroxymethyl-4H-pyran-4-one

In the title compound, $C_6H_6O_5$, an extract from *Gaultheria leucocarpa* B1 var. crenulata, the asymmetric unit consists of two nearly parallel molecules with different conformations linked by a strong intermolecular $O-H\cdots O$ hydrogen bond. The pyranoid ring in both molecules is planar, and the dihedral angle formed by these planes is 7.14 (15)°. The crystal packing is stabilized by $O-H\cdots O$ intermolecular hydrogen bonds, resulting in the formation of a two-dimensional network.

Comment

The title compound, (I), commonly known as 3-hydroxykojic acid, was first isolated from the culture broth of *Gluconobacter cerinus* var. ammoniac (Terada *et al.*, 1961). In the course of our systematic search for bioactive substances from Chinese traditional and herbal medicine, we have studied the leaves of *Gaultheria leucocarpa* B1 var. crenulata, which was widely used for the treatment of rheumatoid arthritis, swelling pain, trauma, chronic tracheitis, cold and vertigo (Jiangsu New Medical College, 1977), and obtained the title compound.



As shown in Fig. 1, the asymmetric unit of (I) consists of two nearly parallel molecules linked by a strong intermolecular $O5-H6\cdots O10$ hydrogen bond. The two molecules differ in the orientation of the hydroxy groups. The pyranoid ring in both molecules is planar, and the dihedral angle between the two planes is 7.14 (15)°.

Although hydroxy groups O2-H2, O4-H3, O7-H8 and O9-H9 are adjacent to the carbonyl C3=O3 and C9=O8 bonds, no significant intramolecular O-H···O hydrogenbond interactions are observed in the crystal structure. The crystal packing is stabilized by strong intermolecular O-H···O hydrogen bonds involving the hydroxy groups and carbonyl atoms O3 and O8, resulting in the formation of a two-dimensional sheet (Fig. 2 and Table 1).

Experimental

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The leaves and stems of *G. leucocarpa* B1 var. crenulata (Kutz, T. Z. Hsu) were collected in the Daming Mountain of Guangxi Province,



Figure 1

An *ORTEP* (Farrugia, 1997; Burnett & Johnson, 1996) view of the asymmetric unit of (I), showing the atom-labeling scheme and 50% probability displacement ellipsoids. The hydrogen bond is shown with dashed lines.

People's Republic of China, and identified by Professor Bing-Yang Ding of Wenzhou Normal University. The air-dried and powdered plant materials (4 kg) were extracted with 95% EtOH. The EtOH extracts were evaporated under reduced pressure to give a residue which was suspended in distilled water and partitioned successively with petroleum ether, CHCl₃, EtOAc and *n*-BuOH. The EtOAC extract (100 g) was introduced into a silica gel column and eluted with CHCl₃–MeOH (10:1) to obtain the title compound (30 mg).

Crystal data

$C_6H_6O_5$	Z = 4
$M_r = 158.11$	$D_x = 1.689 \text{ Mg m}^{-3}$
Triclinic, $P\overline{1}$	Mo $K\alpha$ radiation
a = 8.2045 (9) Å	Cell parameters from 1957
b = 8.9704 (10) Å	reflections
c = 9.6826(10) Å	$\theta = 4.9-56.1^{\circ}$
$\alpha = 93.026 (2)^{\circ}$	$\mu = 0.15 \text{ mm}^{-1}$
$\beta = 109.337 \ (2)^{\circ}$	T = 293 (2) K
$\gamma = 109.787 (2)^{\circ}$	Block, colorless
V = 621.71 (12) Å ³	$0.51 \times 0.50 \times 0.39 \text{ mm}$
Data collection	
Bruker SMART CCD area-detector	2387 independent reflections
diffractometer	2007 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\rm int} = 0.063$
Absorption correction: multi-scan	$\theta_{\rm max} = 26.0^{\circ}$
(SADABS; Sheldrick, 1996)	$h = -8 \rightarrow 10$
$T_{\rm min} = 0.914, T_{\rm max} = 0.939$	$k = -11 \rightarrow 11$
3405 measured reflections	$l = -10 \rightarrow 11$
Refinement	
Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.1098P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.061$	+ 0.1959P]
$wR(F^2) = 0.182$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.08	$(\Delta/\sigma)_{\rm max} < 0.001$
2387 reflections	$\Delta \rho_{\rm max} = 0.55 \text{ e} \text{ Å}^{-3}$
205 parameters	$\Delta \rho_{\rm min} = -0.39 \ {\rm e} \ {\rm \AA}^{-3}$
H-atom parameters constrained	





A CAMERON (Watkin *et al.*, 1993) packing plot of (I), viewed approximately along the crystallographic *a* axis. H atoms not involved in hydrogen bonding have been omitted for clarity and hydrogen bonds are shown as dashed lines.

Table 1

Hydrogen-bonding geometry (Å, °).

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Symmetry codes: (i) 1 + x, y, z; (ii) 2 - x, -y, 1 - z; (iii) 3 - x, 1 - y, 2 - z; (iv) x - 1, y, z.

All H atoms were located in a difference Fourier map, but they were introduced in calculated positions and treated as riding on their parent atoms [C-H = 0.93–0.97 Å, O-H = 0.82 Å, and $U_{iso}(H) = 1.2U_{eq}(C)$ and $1.5U_{eq}(C,O)$].

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT* and *SHELXTL* (Bruker, 2000); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996), *ORTEP-3 for Windows* (Farrugia, 1997) and *CAMERON* (Watkin *et al.*, 1993); software used to prepare material for publication: *SHELXTL*.

This work was supported by a grant from the National Natural Science Foundation of China (30170104). The authors are indebted to Dr Jie Sun of Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, for his valuable suggestion, and Professor Bing-Yang Ding of Wenzhou Normal University for collecting and identifying the plant material.

References

Bruker (2000). *SMART* (Version 6.0), *SAINT* (Version 6.0) and *SHELXTL* (Version 6.10). Bruker AXS Inc., Madison, Wisconsin, USA.

Burnett, M. N. & Johnson, C. K. (1996). ORTEPIII. Report ORNL-6895. Oak Ridge National Laboratory, Tennessee, USA.

Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.

Jiangsu New Medical College (1977). Zhong Yao Da Ci Dian, p. 1879. Shanghai Public Health Publishing House.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Terada, O., Suzuki, S. & Kinoshita, S. (1961). Agric. Biol. Chem. 25, 802–803.
Watkin, D. M., Pearce, L. & Prout, C. K. (1993). CAMERON Chemical Crystallography Laboratory, University of Oxford, England.